

P-Heterocyclic Building Blocks: Application to the Stereoselective Synthesis of *P*-Sugars

Diana S. Stoianova, Alan Whitehead, and Paul R. Hanson*

Department of Chemistry, 1251 Wescoe Hall Drive, Malott Hall, University of Kansas, Lawrence, Kansas 66045-7582

phanson@ku.edu

Received March 14, 2005



A strategy relying on the utilization of stereoselective additions to allyldiphenylphosphonate esters and subsequent ring-closing metathesis (RCM) to access P-chiral P-heterocyclic building blocks for the synthesis of phosphono sugars is described. These building blocks possess several attractive components, including the following: (i) P(2) and C(6) stereogenic centers for directing stereoselective transformations; (ii) an activated C(3) methylene group that promotes base-mediated olefin transposition to generate vinyl phosphonates available for further stereoselective reactions; and (iii) a P(2)-stereogenic center containing an exchangeable phosphonate ester armed to attenuate the "stereochemical environment" at phosphorus. Taken collectively, these attributes contribute to a concise method for the stereoselective synthesis of an array of *P*-sugars.

Carbohydrates offer excellent opportunities to exploit stereochemical diversity, since subtle changes in conformation, stereochemistry, and substituents at various positions are all known to illicit biological responses.¹ Recent interest in the development of new carbohydratebased mimetics has focused on the development of scaffolds that can serve as effective biological probes and/or potential therapeutic agents.¹ A number of innovative synthetic approaches toward this goal have been undertaken,² including the development of new phosphorusbased sugars (P-sugars).³ The emergence of P-sugars has resulted largely from previous successes with organophosphorus compounds,⁴ in both drug⁵ and agricultural⁶ discovery programs. Most notably, recent successes with anomerically modified P-sugars^{3f,3h} have validated their utility and warranted the continued screening of new

derivatives. Overall, a multitude of *P*-sugar analogues have been constructed starting from naturally occurring

^{*} Corresponding author. (1) (a) Gruner, S. A. W.; Locardi, E.; Lohof, E.; Kessler, H. Chem. Rev. 2002, 102, 491-514. (b) Lillelund, V. H.; Jensen, H. H.; Liang, X.; Bols, M. Chem. Rev. 2002, 102, 515-553.
 (2) (a) Kunz, H. Angew. Chem., Int. Ed. Engl. 1987, 26, 294-308.

⁽b) Musser, J. H., Ed. Carbohydrates as Drug Discovery Leads. In Annual Reports in Medical Chemistry; Academic Press: New York, 1992; Vol. 27, pp 301–310. (b) Danishefsky, S. J.; Roberge, J. Y. Pure Appl. Chem. **1995**, 67, 1647–1662.

^{(3) (}a) Thiem, J.; Günther, M.; Paulsen, H.; Kopf, J. Chem. Ber. 1977, 110, 3190-3200. (b) Wróblewski, A. E. Tetrahedron 1986, 42, 3595 3606. (c) Molin, H.; Noren, J. O.; Claesson, A. Carbohydr. Res. 1989, 194, 209-221. (d) Darrow, J. W.; Drueckhammer, D. G. J. Org. Chem. **1994**, 59, 2976–2985. (e) Hanessian, S.; Galeotti, N.; Rosen, P.; Oliva, G.; Babu, S. *Bioorg. Med. Chem. Lett.* **1994**, 4, 2763–2768. (f) Darrow, J. W.; Drueckhammer, D. G. Bioorg. Med. Chem. 1996, 4, 1341-1348. (g) Harvey, T. C.; Simiand, C.; Weiler, L.; Withers, S. G. J. Org. Chem. **1997**, 62, 6722–6725. (h) Hanessian, S.; Rogel, O. Bioorg. Med. Chem. Lett. **1999**, 9, 2441–2446. (i) Hanessian, S.; Rogel, O. J. Org. Chem. 2000, 65, 2667-2674.

^{(4) (}a) Quin, L. D. A Guide to Organophosphorus Chemistry; Wiley-Interscience: New York, 2000. (b) Engel, R. Handbook of Organo-phosphorus Chemistry; Marcel Dekker: New York, 1992. (c) In Phosphorous-Carbon Heterocyclic Chemistry: The Rise of a New Domain; Mathey, F., Ed.; Pergamon: New York, 2001. (d) Quin, L. D. The Heterocyclic Chemistry of Phosphorous: Systems Based on the Phosphorous-Carbon Bond; Wiley & Sons: New York, 1981.

⁽⁵⁾ For examples of phosphorus compounds as pharmaceuticals, see: (a) Kafarski, P.; Lejczak, B. Curr. Med. Chem. 2001, 1, 301-312. (b) Colvin, O. M. Curr. Pharm. Des. 1999, 5, 555-560. (c) Zon, G. Prog.
 Med. Chem. 1982, 19, 205-246. (d) Stee, W. J. Organophosphorus *Chem.* **1982**, *13*, 145–174. (e) Mader, M. M.; Bartlett, P. A. *Chem. Rev.* **1997**, *97*, 1281–1301. (f) Kafarski, P.; Lejczak, B. *Phosphorus, Sulfur,* Silicon Relat. Elem. 1991, 63, 193-215.

⁽⁶⁾ For an example of a phosphorus-containing herbicide, see Franz, J. E.; Mao, M. K.; Sikorski, J. A. *Glyphosate: A Unique Global Herbicide*; American Chemical Society: Washington, DC, 1997.





sugars.³ These routes have mainly employed the Abramov reaction between preformed or commercially available sugar aldehydes and di- or trialkyl phosphites, followed by intramolecular transesterification of the resulting hydroxyphosphonates.^{7,4b} As part of our program aimed at the development of synthetic routes to new *P*-heterocycles with biological and synthetic utility,⁸ we herein report our continued efforts⁹ for the synthesis of novel *P*-sugars.

Our approach hinges on routes emphasizing stereochemical diversity¹⁰ and focuses on the production of a key *P*-chiral allylphostone building block 8, using ringclosing metathesis (RCM),⁹ to generate versatile epimeric vinylphostones 4 and 5, respectively (Scheme 1). These vinylphostone substrates serve as diverse building blocks for the stereo-controlled synthesis of C(3)-C(5) O- and N-substituted P-sugars (1-3, Scheme 1). Several attractive features of scaffold 8 are worth mentioning: (i) P(2)and C(6) stereogenic centers (phostone numbering) for directing stereoselective transformations upon the olefin moiety subunit of 8; (ii) an activated C(3) methylene group that promotes base-mediated olefin transposition to generate vinyl phosphonates 4 and 5 possessing P(2)and C(5) stereocenters for further stereoselective reactions; and (iii) a P(2)-stereogenic center containing an exchangeable phenyl phosphonate ester which allows for attenuation of the "stereochemical environment" at phosphorus. Overall, the method reported herein allows for the generation of an array of stereochemically rich phosphono sugars.

Results and Discussion

Our strategy begins with a stereoselective alkoxide addition of the readily available allylic alcohols **12**, **13**, or **14** to allyl diphenyl phosphonate (**11**) to derive dienes **9** ($R^1 = CH_3$, CH_2OBn , CH_2OTr) (Scheme 2). This





diastereoselective process, first introduced by Moriarty,¹¹ effectively differentiates the two enantiotopic phenyl esters of **11** (ds = 3.5-8:1, selectivities determined by ³¹P NMR). This procedure has been shown to be successful with the commercially available racemic 3-butene-2-ol (**12**) to yield products possessing a 1,3-*trans*-relationship between the PhO- and Me-groups,⁹ or more elaborate allylic alcohols **13** and **14** (vide infra). Subsequent RCM of acyclic precursors **9** using the 1st generation Grubbs metathesis catalyst (cat-**A**)¹² provides allyl phostones **8** in excellent yields.

Elaboration of allylphostone 15, via the aforementioned stereo-divergent oxidation pathways, is outlined in Scheme 3. As previously reported, 9 both C(5) epimers are available from 15 by employing either OsO₄ or trifluoromethyl dioxirane. Dihydroxylation of substrate 15 provided diol **16a** in high stereoselectivity (ds = 15:1) in a process that adheres to the Donohoe/Kishi model,13 whereby dihydroxylation occurs anti to the C(5) hydroxyl group.⁹ Subsequent eliminative opening of the corresponding carbonate or mesylate of the C(4)/C(5) diol subunit in 16a afforded vinyl phostones 17a ($R^1 = H$) and 17b ($R^1 = Ms$). Alternatively, epoxidation of 15 led to the C(5) epimer **16b** in modest selectivity (ds = 4:1), but in excellent isolated yields. As previously reported, we believe the latter is governed by electrostatic repulsion of the P-OR ester and the incoming dipolar dioxirane reagent.¹⁴

⁽⁷⁾ For some recent examples of the synthesis of nonfunctionalized cyclic phosphonates, see: (a) Tasz, M. K.; Rodriguez, O. P.; Cremer, S. E.; Hussain, M. S.; Haque, M. J. Chem. Soc., Perkin Trans. 2 1996, 2221–2226. (b) Yokomatsu, T.; Shioya, Y.; Iwasawa, H.; Shibuya, S. Heterocycles 1997, 46, 463–472. (c) Brel, V. K. Synthesis 1998, 710–712.

⁽⁸⁾ McReynolds, M. D.; Dougherty, J. M.; Hanson, P. R. Chem. Rev. **2004**, *104*, 2239–2258.

 ⁽⁹⁾ Stoianova, D. S.; Hanson, P. R. Org. Lett. 2001, 3, 3285-3288.
 (10) Burke, M. D.; Schreiber, S. L. Angew. Chem., Int. Ed. Engl. 2004, 43, 46-58.

⁽¹¹⁾ Moriarty, R. M.; Tao, A.; Condeiu, C.; Gilardi, R. Tetrahedron Lett. **1997**, 38, 2597–2600.

⁽¹²⁾ Schwab, P.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. **1996**, 118, 8, 100–110.

^{(13) (}a) Cha, J. K.; Christ, W. J.; Kishi, Y. Tetrahedron Lett. 1983,
24, 3943–3946. (b) Cha, J. K.; Christ, W. J.; Kishi, Y. Tetrahedron Lett. 1983, 24, 3947–3950. (c) Donohue, T. J.; Moore, P. R.; Beddoes,
R. L. J. Chem. Soc., Perkin Trans. 1 1997, 43–51.



SCHEME 4



Opening of epoxide 16b with (nucleophilic) Li-alkoxide bases cleanly afforded the desired vinylphostone 17c and also promoted phenoxy displacement with retention of configuration at phosphorus in addition to the eliminative opening. The relative stereochemistry of compounds 16b and 17d ($R^2 = Me$) was determined by X-ray crystallographic analysis.⁹ It is noteworthy that attempts to open the epoxide with nonnucleophilic bases such as KHMDS or LDA resulted in decomposition.

Functionalization of vinyl phostone intermediate 17a is highlighted in Scheme 4. Directed dihydroxylation using the Donohoe protocol¹⁵ with stoichiometric amounts of OsO₄/TMEDA provided the C(3)-C(5) all syn-triol 18a with excellent selectivity (ds = 13:1). Hydrolysis of the osmate ester was accomplished with citric acid, which has been previously shown to facilitate catalytic osmiumpromoted dihydroxylation reactions due to its ability to readily hydrolyze osmate esters.¹⁶ The excellent selectivity of this directed-dihydroxylation reaction is in agreement with the results of Donahoe, showing that substrates possessing equatorial hydroxy-groups are very selective under hydrogen-bonding conditions.¹⁷

Stoianova et al.





Dihydroxylation of 17a under catalytic OsO4 conditions was previously shown, by X-ray crystallographic analysis of the product, to occur anti to the C(5) allylic hydroxy group to afford triol **18b** in modest selectivity (ds = 5:1).⁹ We were able to improve this selectivity by protecting the hydroxy-group as a benzyl ether using in situ formed benzyl triflate to derive 19. Subsequent dihydroxylation generated triol 20 in excellent yield (92%) and selectivity $(15:1).^{18}$

We assumed that the dihydroxylation of 19 occurred anti to the C(5) benzyloxy group to derive 20, in an analogous manner to the dihydroxylation of 17a affording 18b.¹⁹ This assumption was confirmed by three-bond coupling analysis of coupling constants $(J_{\rm HP})$ between the C(4) proton and P(1) atoms in the fully dihydroxylated compounds 20 and 18a as shown in Figure 1. A significantly higher coupling constant (35 Hz) is observed for 18a when the H-C-C-P dihedral angle is 180°. Alternatively, an H-C-C-P dihedral angle of <90° in 20 affords a much smaller coupling constant (5.8 Hz). This coupling constant data was substantiated by a 5.8 Hz coupling in the previously reported X-ray supported structure 18b.19

Alternatively, introduction of nitrogen at the C(4)position could also be achieved from vinylphosphonate 17a via treatment with Cl₃CCN to form the corresponding trichloroacetimidate, which underwent facile 1,4conjugate addition to afford 21 in excellent yield. Attempts at the aminohydroxylation of 17a have thus far failed.

Introduction of nitrogen nucleophiles at the C(5) position was achieved from the mesylated vinylphosphonate **17b.** Thus, treatment with sodium azide afforded C(5)inverted azide product 22. Subsequent dihydroxylation proceeded with complete selectivity to afford 23 as a single diastereomer (Scheme 5). The relative stereochem-

^{(14) (}a) D'Accolti, L.; Fiorentino, M.; Fusco, C.; Rosa, A. M.; Curci, R. Tetrahedron Lett. 1999, 40, 8023-8027. (b) Curci, R.; Dinoi, A.; Rubino, M. F. Pure Appl. Chem. 1995, 67, 811-822. (c) Adam, W.; Paredes, R.; Smerz, A. K.; Veloza, L. A. Liebigs Ann. 1997, 547-551. (15) Donohoe, T. J. Synlett 2002, 1223-2332.

⁽¹⁶⁾ Dupau, P.; Epple, R.; Thomas, A. A.; Fokin, V. A.; Sharpless, B. K. Adv. Synth. Catal. 2002, 344, 421-433.

⁽¹⁷⁾ Donohoe, T. J.; Blades, K.; Helliwell, M.; Moore, P. R.; Winter, J. J. J. Org. Chem. 1999, 64, 2980-2981.

⁽¹⁸⁾ Benzylation with BnBr under basic conditions gave very low yields of the desired product accompanied by some epimerization. For use of BnBr/AgOTf for benzylation of alcohols, see: Burk, R. M.; Gac, T. S.; Roof, M. B. Tetrahedron Lett. 1994, 44, 8111-8112

⁽¹⁹⁾ We have previously reported the X-ray structure of 18b; see ref 9.





istry of this compound was confirmed by X-ray crystallographic analysis.²⁰

Having established this method with a racemic allylic alcohol, we turned to the use of nonracemic allyl alcohols, which we conveniently derived from commercially inexpensive enantio-enriched (97-99% ee) benzyl- and tritylprotected²¹ glycidyl ethers.^{22,23} Transformation of these readily available epoxides into corresponding allyl alcohols 13 and 14 was achieved in a single step via addition of trimethylsulfonium ylide (Scheme 6).24 Diastereoselective addition of benzyl protected (2S)-1,2-butenediol 13 proceeded in good yield and selectivity (ds = 6:1) to produce phosphonate 24 as a mixture of P(2)-epimers. RCM of 24 provided nonracemic allyl phostone intermediate 25. Dihydroxylation of substrate 25 provided diol **26** as a single diastereomer using 1% OsO₄, *m*-CPBA and NMM (N-methyl morpholine). Noteworthy, reactions with 4% OsO₄, *m*-CPBA/NMM or OsO₄/NMO gave a dramatic decrease in selectivity (ds = 2-5:1). Subsequent carbonate formation using the aforementioned triphosgene conditions, and elimination led to vinylphosphonate (-)-27 in good yield.

Installation of nitrogen nucleophiles at the C(5) position within the framework of nonracemic benzyl-protect-

JOC Article



ed alcohol **26** was achieved using similar sequences as previously described for **17b** (Scheme 7). Formation of the mesylated diol intermediate followed by elimination provided vinyl phostone **28**. Addition of a single equivalent of azide produced **29** in excellent yield.²⁵ Dihydroxylation of the vinyl phostone **29** produced azido *P*-sugar (+)-**30** as a single diastereoisomer in modest yield.

Alternatively, installation of nitrogen at the C(4) position can be accomplished by facile formation of the trichloroacetimidate and subsequent Michael addition in the presence of DBU to yield the bicyclic trichloroacacoline (Scheme 8). Hydrolysis of this product yielded the trichloroacetamide of the C(4)/C(5) amino alcohol. Dihydroxylation of **27** gave triol **31** with excellent selectivity (ds = 15:1), but only moderate yield (53%).

Problems associated with the base lability of phenoxy phosphonate esters led us to develop an alternate route employing BnO-substituted phosphonate esters. Thus, mixed phosphonate ester **34** could be produced by displacement of the PhO-group in the acyclic mixed phosphonate with BnOLi in DME (Scheme 9). This reaction occurs with inversion of the configuration at phosphorus²⁶ to afford **34**.²⁷ Subsequent RCM afforded phostone **35** possessing a cis relationship between the C(6) trityloxymethyl group and the P(2)-benzyl phosphonate ester. Dihydroxylation of **35** with OsO₄/NMO in the presence of citric acid yielded diol **36** with excellent selectivity (*ds*

⁽²⁰⁾ See Supporting Information (S-68).

⁽²¹⁾ The mixed phosphonate, derived from optically pure Trprotected (2S)-1,2-butenediol, was also prepared as described in our previous paper. Stability issues with the Tr-protected compounds led us to focus on the more robust Bn-protected vinylphosphonate (+)-27.

⁽²²⁾ We kindly acknowledge Daiso Co., Ltd., Fine Chemical Department for donating 100 g of each antipode of both the benzyl and trityl protected glycidols (e-mail: akkimura@daiso.co.jp).

⁽²³⁾ Schaus, S. E.; Brandes, B. D.; Larrow, J. F.; Tokunaga, M.; Hansen, K. B.; Gould, A. E.; Furrow, M. E.; Jacobsen, E. N. J. Am. Chem. Soc. **2002**, 124, 1307–1315.

⁽²⁴⁾ Davoille, R. J.; Rutherford, D. T.; Christie, S. D. R. Tetrahedron Lett. 2000, 41, 1255–1259.

⁽²⁵⁾ Displacement of the C(5) mesylate in compound **28** using a slight excess of sodium azide was complicated by the formation of an inseparable side product (5–10 mol %). This side product most likely arose from conjugative addition of a second azide into the vinylphosphonate.

⁽²⁶⁾ Thatcher, G. R. J.; Kluger, R. Adv. Phys. Org. Chem. **1989**, 25. (27) Product **34** was contaminated with $\sim 4\%$ of the vinylphosphonate arising from double bond isomerization. Reactions run in THF gave extensive isomerization of the double bond (40%).



= 10-14:1) and in good isolated yields. On the basis of the aforementioned selectivity seen in the dihydroxylation of the C(6)-Me-substituted allylphostone 15, we tentatively assigned the major product 36 occurring via dihydroxylation anti to the C(6)-trityloxymethyl group.²⁸ Conversion of diol 36 into the corresponding carbonate and treatment with KHMDS gave vinyl phosphonate 37 (Scheme 9). Dihydroxylation of this compound under standard conditions (cat. OsO4, NMO, citric acid) proceeded with very low selectivity (ds = 2:1), presumably due to the anti relationship between the P(2)-benzyloxy phosphonate ester and the equatorial C(5)-OH which provide opposing factors. Alternatively, directed-dihydroxylation using the Donohoe conditions¹⁵ provided C(3)/ C(4)/C(5) all syn-triol 38 as a single diastereoisomer (ds > 20:1). Attempts to hydrolyze the osmate ester using literature conditions (HCl in MeOH)⁸ gave only decomposition of the product. Treatment with citric acid¹⁶ gave clean hydrolysis of the osmate without cleaving the Trgroup. This product was filtered through silica gel²⁹ and treated with catalytic amount of TsOH to give fully deprotected tetraol 38.

In our previous work outlined in Scheme 3, we used epoxidation to derive 16b as an alternative route of functionalizing the RCM product 15, and ultimately to derive diastereomer 17c. Unlike the dihydroxylation of 15 (Scheme 3), the major product of the epoxidation resulted from attack syn to the C(6)-methyl group. Unfortunately, all epoxidations of nonracemic phostones 25 and 35 were low yielding and not very selective. Attempts to access a vinyl phosphonate with a syn stereochemical relationship between the C(5) and C(6)substituents using Mitsunobu inversion of the C(5)hydroxy group in 37 gave only starting material. Alternatively, the desired intermediate was accessed by conversion of diol 36 into mesylate 39 and carrying out S_N2-substitution with an alkoxide-containing nucleophile. The best results were seen using KOBz in the presence





of 18-Crown-6 in DMF. Use of other reagents (Bu_4NOAc , NaOAc, KO_2 , $CH_3CH_2CO_2Cs/CH_3CH_2CO_2H$) led to cleavage of the Bn-group on the P-atom and formation of a substantial amount of the free acid.³⁰ Dihydroxylation of **40** gave diol **42** in good yields and selectivity, but attempts to deprotect this compound were unsuccessful. Deprotection of the benzoate in **40** using Mg(OMe)₂, followed by dihydroxylation, and cleavage of the Tr-group gave desired tetraol **43** (Scheme 10) as a single diastereoisomer.

In conclusion, a facile strategy has been developed that allows for the diastereoselective generation of a number of novel *P*-sugar analogues. Additional stereoselective routes to nonracemic phosphono sugars are currently being pursued as well as the generation of libraries of pertinent analogues. Further synthetic studies, the production of chemical libraries, and biological screening of these *P*-sugars are underway and will be reported in due course.

Experimental Section

Experimental Procedures and Spectral Data for Compounds 17b, 18–42.

 $(2S_P^*, 3S^*, 4R^*, 5R^*, 6R^*)$ -6-Methyl-2-phenoxy-1,2-oxaphosphorinane-3,4,5-triol-2-oxide [(±)-18a]. To a solution of 17a (20 mg, 0.083 mmol) in CH_2Cl_2 (1.5 mL) were added OsO_4 (28 mg, 0.11 mmol) and TMEDA (13 mg, 0.11 mmol) at -60 °C. The reaction mixture was stirred for 2 h at -60 °C and warmed to room temperature. Removal of the solvent under reduced pressure under reduced pressure afforded the crude osmate as a mixture of diastereoisomers (12.5:1 by $^{31}\!\mathrm{P}$ NMR analysis). The product was dissolved in 1.5 mL MeOH and treated with citric acid overnight. Removal of the solvent under reduced pressure and flash chromatography (EtOAc/4% MeOH) afforded 16 mg (70%) of **18a** as an oil; ¹H NMR δ 7.29–7.09 (m, 5H), 4.63-4.55 (m, 1H), 4.26 (dd, J = 35.2, 3.9, 2.5 Hz, 1H), 4.05 (dd, J = 7.6, 4.0 Hz, 1 H), 3.38 (dd, J = 9.4, 2.2 Hz, 1 H),1.38 (dd, $J=6.3,\,1.9$ Hz, 3H); $^{13}{\rm C}$ NMR δ 150.0 (d, $J_{\rm CP}=10.0$ Hz), 129.5 (d, $J_{\rm CP} = 125.0$ Hz), 120.3 (d, $J_{\rm CP} = 4.3$ Hz), 74.4 (d, $J_{\rm CP} = 6.2$ Hz), 72.5, 72.3, 66.3 (d, $J_{\rm CP} = 146.9$ Hz), 18.4 (d, $J_{\rm CP}$ = 8.3 Hz); ³¹P NMR δ 16.4.

 $(2S_{P}\text{*},5S^{*},6R^{*})\text{-}5,6\text{-}Dihydro-6\text{-}methyl-2\text{-}phenoxy-2H-1,2-}oxaphosphorinan e-5-methanesulfonate-2-oxide [(±)-17b]. To a solution of the diol 16a (20 mg, 0.077 mmol) in CH_2Cl_2 (2.0 mL) was added a catalytic amount of DMAP (1.9 mg, 0.015 mmol), MsCl (0.02 mL, 0.26 mmol), and Et_3N (0.04$

⁽²⁸⁾ Exhaustive deprotection of compounds 26 and 36 gave rise to a common free acid, thus confirming that diols 26 and 36 differ only in their substitution at phosphorus.

 $[\]left(29\right)$ Attempts to isolate this product led to lower yields due to its instability.

⁽³⁰⁾ Treatment of the phenoxy substituted P-sugar **28** (Scheme 7) with KOBz, 18-Crown-6 in DMF gave almost exclusively the free acid.

mL, 0.29 mmol) at 0 °C. The reaction mixture was warmed to room temperature over 1 h and stirred overnight. Brine was added, and the mixture was extracted with EtOAc (3x) and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (7:3 EtOAc/hexanes) afforded 19 mg (78%) of **17b**; IR (neat) 3028, 2936, 1591, 1490, 1455, 1361, 1270, 1177, 1070, 1023, 931 cm⁻¹; ¹H NMR δ 7.38–7.18 (m, 5H), 6.82 (ddd, J = 46.8, 12.9, 2.0 Hz, 1H), 6.26 (ddd, J = 17.4, 12.9, 2.0 Hz, 1H), 5.17–5.13 (m, 1H), 4.75–4.70 (m, 1H), 3.12 (s, 3H), 1.57 (dd, J = 6.3, 2.0 Hz, 3H); ¹³C NMR δ 149.6, 146.1, 129.9, 125.4, 120.3, 119.7 (d, $J_{CP} = 169.7$ Hz), 76.3 (d, $J_{CP} = 8.1$ Hz), 75.1 (d, $J_{CP} = 10.8$ Hz), 38.8, 18.9 (d, $J_{CP} = 6.9$ Hz); ³¹P NMR δ 5.5. Exact mass: calcd for C₁₂H₁₆O₆PS (M+1) 319.0405; found 319.0405 (FAB).

(2S_P*,5S*,6R*)-5,6-Dihydro-6-methyl-2-phenoxy-2H-1,2oxaphosphorinan e-5-phenylmethoxy-2-oxide $[(\pm)-19]$. To a solution of the alcohol 17a (10 mg, 0.042 mmol) in CH₂Cl₂ (0.6 mL) were added 2,6-di-*t*-butyl-4-methyl pyridine (55 mg, 0.26 mmol), AgOTf (54 mg, 0.21 mmol), and BnBr (0.03 mL, 0.25 mmol) at 0 °C. The reaction mixture was stirred at the same temperature for 3 h and diluted with CH₂Cl₂, and the filtrate was washed with brine. Removal of the solvent under reduced pressure and flash chromatography (4:1 hexanes/EtOAc) afforded 10 mg (72%) of 19 as an oil; IR (neat) 3061, 3031, 2982, 2934, 2871, 1723, 1593, 1490, 1454, 1364, 1270, 1202, 1095, 1067, 982, 928 cm $^{-1}$; ¹H NMR δ 7.41–7.31 (m, 7H), 7.21-7.15 (m, 3H), 6.83 (ddd, J = 48.2, 12.9, 1.6 Hz, 1H), 6.10 (ddd, J = 18.8, 12.9, 2.1 Hz, 1H), 4.72–4.57 (m, 3H), 4.03–3.99 (m, 1H), 1.49 (dd, J = 6.2, 2.0 Hz, 3H); ¹³C NMR δ 149.6, 136.6, 129.8, 128.7, 128.5, 125.1, 120.5 (d, $J_{\rm CP} = 4.6$ Hz), 116.8 (d, $J_{\rm CP} = 170.8$ Hz), 77.9 (d, $J_{\rm CP} = 8.2$ Hz), 76.5 (d, $J_{\rm CP} =$ 10.1 Hz), 72.8, 19.1 (d, $J_{CP} = 7.5$ Hz); ³¹P NMR δ 7.7. Exact mass: calcd for $C_{18}H_{20}O_4P$ (M+1) 331.1099; found 331.1098 (FAB).

 $(2S_P^*, 3R^*, 4S^*, 5S^*, 6R^*)$ -6-Methyl-2-phenoxy-1,2-oxaphosphorinane-3,4-diol-5-phenylmethoxy-2-oxid $e[(\pm)-20]$. To a solution of 19 (15 mg, 0.045 mmol) in acetone (0.9 mL) and t-BuOH (0.3 mL) were added citric acid (19 mg, 0.10 mmol), NMO·H₂O (12 mg, 0.09 mmol), and OsO₄ (2 drops of a 4%solution in water). The reaction mixture was stirred for 24 h, followed by the addition of Na₂SO₃. Water was added, and the mixture was extracted with EtOAc (3x) and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (EtOAc/1% MeOH) afforded 15 mg (92%) of **20** as an oil; ¹H NMR δ 7.36–7.16 (m, 10H), 5.45 (br. s, 1H), 4.94 (d, J = 10.7 Hz, 1H), 4.65 (t, J = 10.8 Hz, 1H), 4.38 (q, J)= 3.8 Hz, 1H), 4.22–4.11 (m, 2H), 3.62 (t, J = 9.2, 1H), 3.24 (d, J = 8.6, 1.6 Hz, 1H), 1.47 (dd, J = 6.2, 1.3 Hz, 3H); ¹³C NMR δ 149.6, 137.6, 130.0, 128.5, 128.2, 128.1, 125.4, 119.7 (d, $J_{\rm CP}$ = 4.7 Hz), 80.0, 75.7, 75.5 (d, $J_{\rm CP}$ = 5.8 Hz), 73.5 (d, $J_{\rm CP}$ = 6.2 Hz), 65.6 (d, $J_{\rm CP}$ = 144.3 Hz), 19.2 (d, $J_{\rm CP}$ = 9.0 Hz); ³¹P NMR δ 16.2.

[(±)-21]. To a solution of 17a (16 mg, 0.067 mmol) in CH₂Cl₂ (0.6 mL) were added Cl₃CCN (0.01 mL) and a catalytic amount of DBU (2.0 mg, 0.013 mmol) at -50 °C, and the solution was warmed to -20 °C over 4 h. The cooling bath was removed, and the stirring was continued at room temperature overnight. The reaction mixture was quenched with NH₄Cl (sat'd aq), extracted with CH₂Cl₂ (3x) and dried (Na₂- SO_4). Removal of the solvent under reduced pressure and flash chromatography (EtOAc/1% MeOH) afforded 24 mg (93%) of product 21; IR (neat) 2918, 1725, 1521, 1486, 1256, 1182, 1039, 1008, 918, 899 cm $^{-1}$; ¹H NMR δ 7.37–7.18 (m, 5H), 4.97–4.88 (m, 1H), 4.83 (d, J = 9.0 Hz, 1H), 4.49 - 4.44 (m, 1H), 2.80 (ddd, J)J = 18.3, 15.6, 6.5 Hz, 1H), 2.29 (ddd, J = 17.8, 15.8, 9.5 Hz, 1H), 1.59 (dd, J = 6.2, 1.0 Hz, 3H); $^{13}\mathrm{C}$ NMR δ 162.4 (d, J_{CP} = 6.6 Hz), 149.4 (d, $J_{\rm CP}=$ 8.9 Hz), 129.9, 125.4, 119.9 (d, $J_{\rm CP}=$ 4.8 Hz), 77.1 (d, $J_{CP} = 6.6$ Hz), 70.4 (d $J_{CP} = 5.5$ Hz), 48.8 (d, $J_{\rm CP} = 4.0$ Hz), 24.6 (d, $J_{\rm CP} = 126.9$ Hz), 19.0 (d, $J_{\rm CP} = 6.2$ Hz); $^{31}\mathrm{P}$ NMR δ 20.0.

(2S_P*,5*R**,6*R**)-5,6-Dihydro-6-methyl-2-phenoxy-2H-1,2-oxaphosphorinan e-5-azido-2-oxide [(±)-22]. To a solution of mesylate **17b** (13 mg, 0.041 mmol) in DMF (0.5 mL) were added NaN₃ (13 mg, 0.20 mmol) and 15-crown-5 (9 mg, 0.041 mmol). The reaction mixture was stirred overnight at room temperature, and the solvent was removed under reduced pressure. Flash chromatography of the residue (1:1 hexanes/EtOAc) afforded 10 mg (92%) of **22**; IR (neat) 3049, 2993, 2938, 2109, 1612, 1596, 1490, 1455, 1358, 1269, 1162, 1028, 968, 921 cm⁻¹; ¹H NMR δ 7.36–7.17 (m, 5H), 6.85 (ddd, J = 46.6, 12.5, 55 Hz, 1H), 6.44 (dd, J = 17.3, 12.5 Hz, 1H), 4.92–4.89 (m, 1H), 3.56–3.55 (m, 1H), 1.54 (dd, J = 6.4, 2.0 Hz, 3H); ¹³C NMR δ 149.8, 142.4, 129.9, 125.2, 121.7 (d, $J_{\rm CP} = 169.5$ Hz), 120.2 (d, $J_{\rm CP} = 4.6$ Hz), 76.8 (d, $J_{\rm CP} = 7.5$ Hz), 57.1 (d, $J_{\rm CP} = 10.1$ Hz), 18.5 (d, $J_{\rm CP} = 7.8$ Hz); ³¹P NMR δ 5.8. Exact mass: calcd for C₁₁H₁₃N₃O₃P (M+1) 266.0695; found 266.0692 (FAB).

 $(2S_P^*, 3S^*, 4R^*, 5R^*, 6R^*)$ -6-Methyl-2-phenoxy-1,2-oxaphosphorinane-3,4-diol-5-azido-2-oxide [(\pm) -23]. To a solution of 22 (5 mg, 0.019 mmol) in acetone (0.6 mL) and t-BuOH (0.2 mL) were added citric acid (4 mg, 0.021 mmol), NMO·H₂O (3 mg, 0.022 mmol), and OsO_4 (1 drop 4% solution in water). The reaction mixture was stirred for 24 h, and water was added and extracted with EtOAc (3x). Removal of the solvent under reduced pressure and flash chromatography (4:1 EtOAc/ hexanes) afforded 5 mg (88%) of 23; IR (neat) 3350, 2923, 2110, 1590, 1491, 1255, 1199, 1026, 987 cm $^{-1};$ $^1\mathrm{H}$ NMR δ 7.33 – 7.16 (m, 5H), 5.84 (br. s, 1H), 5.10–5.04 (m, 1H), 4.40 (dt, J = 35.2, 4.2 Hz, 1H), 4.25 (t, J = 4.5 Hz, 1H), 3.42 (br. s, 1H), 3.73 (d, J=4.2 Hz, 1H), 1.47 (dd, J=6.6, 1.6 Hz, 3H); $^{13}\mathrm{C}$ NMR δ 150.2 (d, $J_{\rm CP} = 9.8$ Hz), 129.7, 125.1, 120.4 (d, $J_{\rm CP} = 4.3$ Hz), $73.4 (d, J_{CP} = 7.1 Hz), 71.8 (d, J_{CP} = 2.9 Hz), 65.2 (d, J_{CP} = 2.6 Hz)$ Hz), 63.7 (d, $J_{CP} = 145.7$ Hz), 18.0 (d, $J_{CP} = 9.0$ Hz); ³¹P NMR δ 15.9. Exact mass: calculated for $C_{11}H_{15}N_3O_5P\left(M{+}1\right)$ 300.0749; found 300.0754 (FAB).

(1S,S_P)-2-Propenyl-1-phenylmethoxymethyl-2-propenyl Phenyl Ester Phosphonic Acid [(+)-24]. To a solution of the 2-benzyl-(2R)-1,2-butenediol 13 (69 mg, 0.388 mmol) in dry THF (2 mL) was added BuLi (0.27 mL 1.4 M in hexanes, 0.38 mmol) at -40 °C, and the solution was warmed to room temperature over 2 h and stirred at room temperature for an additional 4 h. This solution was added to the diphenyl allylphosphonate (127 mg, 0.42 mmol) in THF (0.6 mL) and HMPA (0.1 mL) at -78 °C over 30 min. The reaction mixture was stirred overnight at -78 °C, quenched with NH₄Cl (sat'd aq.), extracted with EtOAc (3x), and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (4:1 hexanes/EtOAc) afforded 75 mg (54%) of 24 as a mixture of diastereoisomers (6:1 by ³¹P analysis); $[\alpha]_D$ +5.3 (c 0.3, CH₂Cl₂); IR (neat) 3065, 3029, 2982, 2861, 1639, 1593, 1490, 1454, 1364, 1265, 1208, 1091, 1013, 926 cm⁻¹; ¹H NMR (only signals for major isomer) δ 7.36–7.11 (m, 10H), 5.92– 5.14 (m, 2H), 5.46-5.14 (m, 5H), 4.64-4.53 (m, 2H), 3.62-3.53 (m, 2H), 2.85-2.75 (m, 2H); ¹³C NMR (only signals for major isomer) δ 150.3 (d, $J_{CP} = 8.9$ Hz), 137.5, 129.4, 128.3, 127.7, 127.6, 124.7, 124.63, 120.59, 120.2 (d, $J_{CP} = 15.1$ Hz), 118.3, 76.5 (d, $J_{\rm CP}$ = 6.6 Hz), 73.0, 72.3 (d, $J_{\rm CP}$ = 4.8 Hz), 32.1 (d, $J_{\rm CP}$ = 140.7 Hz); ³¹P NMR δ 25.6. Exact mass: calculated for C₂₀H₂₄O₄P (M+1) 359.1412; found 359.1406 (FAB)

(2S_P,6S)-2-Phenoxy-6-[(phenylmethoxymethyl]-2H-1,2oxaphosphorin-3,6-dihydro-2-oxide [(-)-25]. To a solution of 24 (75 mg, 0.23 mmol, 9:1 mixture) in dry CH₂Cl₂ (20 mL) was added (PCy₃)₂Cl₂Ru=CHPh (6 mg, 3 mol %), and the mixture was refluxed overnight. An additional amount of catalyst (3 mg, 0.0036 mmol) was added, and reflux was continued for 14 h. The mixture was diluted with CH₂Cl₂ and flushed with air, 3 drops DMSO were added, and the solution was stirred overnight at room temperature. Removal of the solvent under reduced pressure and flash chromatography (1:1 hexanes/EtOAc) afforded 60 mg (83%) of 25 as an oil; $[\alpha]_D$ -42.1 (*c* 1.2, CH₂Cl₂); IR (neat) 3062, 3033, 2905, 2866, 1592, 1491, 1454, 1297, 1254, 1207, 1083, 1026, 927 cm⁻¹; ¹H NMR δ 7.36-7.13 (m, 10H), 5.88-5.77 (m, 2H), 5.14-5.13 (m, 1H), 4.58 (s, 2H), 3.73-3.62 (m, 2H), 2.68-2.48 (m, 2H); ¹³C NMR δ 149.9 (d, $J_{\rm CP}$ = 7.8 Hz), 137.4, 129.7, 128.2, 127.6, 127.5, 126.2 (d, $J_{\rm CP}$ = 16.0 Hz), 124.8, 121.2 (d, $J_{\rm CP}$ = 9.8 Hz), 129.9 (d, $J_{\rm CP}$ = 4.5 Hz), 79.7 (d, $J_{\rm CP}$ = 8.5 Hz), 63.4, 71.6 (d, $J_{\rm CP}$ = 7.4 Hz), 21.4 (d, $J_{\rm CP}$ = 132.4 Hz); $^{31}{\rm P}$ NMR δ 17.5. Exact mass: calculated for $\rm C_{18}H_{20}O_4P$ (M+1) 331.1099; found 331.1110 (FAB).

(2Sp,4R,5S,6S)-2-Phenoxy-6-[(phenylmethoxy)methyl]-1,2-oxaphosphorinane-4,5-diol-2-oxide [(+)-26]. To a solution of 25 (12 mg, 0.036 mmol in 1.2 mL acetone and t-BuOH (0.4 mL) were added citric acid (10 mg, 0.052 mmol), m-CPBA (26 mg 70-75%), NMM (11 mg, 0.11 mmol) and OsO₄ (1 drop of a 1% solution in water). After stirring for 24 h one more drop of OsO₄ was added and the stirring continued for 24 h. The reaction mixture was diluted with CHCl₃, solid Na₂SO₃ and water was added, stirred for 30 min and extracted with CHCl₃. Removal of the solvent under reduced pressure and flash chromatography (EtOAc/1% MeOH) afforded 11 mg (84%) of **26** as a white solid; mp = 145-148; $[\alpha]_D + 51.1$ (c 0.980, CH₂Cl₂); IR (neat) 3341, 3062, 2969, 1591, 1490, 1454, 1251, 1197, 933 cm $^{-1};\,^1\!\mathrm{H}$ NMR δ 7.32–7.11 (m, 10H), 4.71–4.68 (m, 1H), 4.55-4.48 (m, 2H), 4.12 (dd, J = 35.8, 1.8 Hz, 1H), 3.85(br. s, 1H), 3.75-3.71 (m, 4H), 2.32 (ddd, J = 19.4, 15.8, 3.6 Hz, 1H), 2.04–1.94 (m, 1H); ¹³C NMR δ 149.6 (d, $J_{CP} = 8.4$ Hz), 137.6, 129.9, 129.6, 128.3, 127.7, 127.5, 125.1, 120.6 (d, $J_{\rm CP} = 4.3$ Hz), 76.4 (d, $J_{\rm CP} = 6.7$ Hz), 73,4, 69.9 (d, $J_{\rm CP} = 7.8$ Hz), 68.0 (d, $J_{\rm CP} = 6.5$ Hz), 67.1 (d, $J_{\rm CP} = 9.1$ Hz), 27.2 (d, $J_{\rm CP}$ = 125.9 Hz); ³¹P NMR δ 20.2. Exact mass: calculated for C₁₈H₂₂O₆P (M+1) 365.1154; found 365.1160 (FAB).

(2S_P,4R,5S,6S)-2-Phenoxy-6-[(phenylmethoxy)methyl]-1,2-oxaphosphorinane-4,5-carbonate-2-oxide. To a solution of 26 (450 mg, 1.24 mmol) in CH₂Cl₂ (20 mL) were added triphosgene (770 mg, 2.59 mmol) and Et₃N (0.80 mL, 5.75 mmol) at -40 °C, and the solution was warmed to 10 °C over 3 h. The reaction mixture was diluted with CH₂Cl₂, quenched with NH₄Cl (sat'd aq.), extracted with CH₂Cl₂ (3x), and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (7:3 EtOAc/hexanes) afforded 460 mg (95%) of the carbonate; ¹H NMR δ 7.42–7.18 (m, 10H), 5.25 (ddd, J = 25.0, 7.8, 6.2, 4.1 Hz, 1H), 4.94 (dd, J = 8.1, 4.5 Hz)1H), 4.86–4.85 (m, 1H), 4.65 (d, J = 11.1 Hz, 1H), 4.53 (d, J = 11.1 = 11.5 Hz, 1H), 3.94 (dd, J = 11.1, 2.0 Hz, 1H), 3.80 (dt, J =11.1, 3.0 Hz, 1H), 2.70 (ddd, J = 18.7, 16.3, 6.2 Hz, 1H), 2.48 (ddd, J= 17.6, 16.3, 4.0 Hz, 1H); $^{13}\mathrm{C}$ NMR δ 152.6, 149.3 (d, $J_{\rm CP} = 8.6$ Hz), 136.4, 129.6, 128.4, 128.0, 127.7, 125.2, 120.1 (d, $J_{\rm CP} = 4.7$ Hz), 77.9 (d, $J_{\rm CP} = 7.7$ Hz), 73,6, 73.0 (d, $J_{\rm CP} =$ 6.1 Hz), 72.7 (d, $J_{CP} = 11.9$ Hz), 69.6 (d, $J_{CP} = 3.8$ Hz), 24.3 (d, $J_{\rm CP} = 128.2$ Hz); ³¹P NMR δ 15.2.

(2SP,5S,6R)-5,6-Dihydro-2-phenoxy-6-[(phenylmethoxy)methyl]-2H-1,2-oxaphosphorinane-5-ol-2-oxide [(-)-27]. To a solution of the carbonate (430 mg, 1.10 mmol) in THF (20 mL) was added KHMDS (2.5 mL, 0.5 M in toluene, 1.25 mmol) at -50 °C, and the solution was warmed to 0 °C over 2.5 h. The reaction mixture was quenched with NH₄Cl (sat'd aq.), extracted with EtOAc (3x), and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (9:1 EtOAc/hexanes) afforded 280 mg (74%) of 27; [a]_D -4.6 (c 0.5, CH₂Cl₂); IR (neat) 3341, 3062, 2869, 1591, 1490, 1454, 1356, 1251, 1198, 1077, 1007, 985, 932 cm $^{-1};$ $^1{\rm H}$ NMR δ 7.34-7.12 (m, 10H), 6.68 (ddd, J = 49.4, 12.9, 1.3 Hz, 1H), 5.93-5.85 (m, 1H), 4.70-4.49 (m, 4H), 4.36 (d, J = 1.5 Hz, 1H), 3.83-3.71 (m, 1H); ¹³C NMR δ 154.0, 149.5 (d, $J_{CP} = 8.5$ Hz), 137.5, 129.1, 128.3, 127.7, 127.5, 125.3, 120.5 (d, $J_{\rm CP}=$ 4.2 Hz), 114.4 (d, $J_{CP} = 171.7$ Hz), 81.4, 73.4, 69.7 (d, $J_{CP} =$ 7.5 Hz), 64.3; ³¹P NMR δ 8.0. Exact mass: calculated for C₁₈H₂₀O₅P (M+1) 347.1048; found 347.1055 (FAB).

 $(2S_P, 5S, 6R)$ -5,6-Dihydro-2-phenylmethoxy-2H-1,2oxaphosphorinane-5-methanesulfonate-6-[(triphenylmethoxy)methyl]-2-oxide [(+)-28]. To a solution of the diol (100 mg, 0.27 mmol) in CH₂Cl₂ (8 mL) were added a catalytic amount of DMAP (6.6 mg, 0.054 mmol) and MsCl (0.07 mL, 0.90 mmol) at 0 °C, followed by Et₃N (0.15 mL, 1.08 mmol). The reaction mixture was warmed to room temperature over

1 h and stirred overnight at room temperature. Et₃N (0.10 mL, 0.72 mmol) was added, and the stirring was continued for 6 h. After addition of brine the mixture was extracted with EtOAc (3x) and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (1:1 hexanes/EtOAc) afforded 95 mg (83%) of **28**; $[\alpha]_{\rm D}$ +30.7 (c 0.91, CH₂Cl₂); IR (neat) 3060, 3044, 3023, 2931, 2896, 2865, 1590, $1491, 1454, 1348, 1268, 1204, 1171, 1101, 1018, 985, 939 \text{ cm}^{-1}$ ¹H NMR δ 7.37–7.16 (m, 10H), 6.87 (ddd, $J=47.2,\,12.9,\,1.9$ Hz, 1H), 6.24 (ddd, J = 17.9, 12.9, 2.0 Hz, 1H), 5.53–5.49 (m, 1H), 4.76-4.71 (m, 1H), 4.66 (d, J = 11.5 Hz, 1H), 4.60 (d, J= 11.7 Hz, 1H), 3.88–3.80 (m, 2H), 3.00 (s, 3H); $^{13}\mathrm{C}$ NMR δ $149.5 (d, J_{CP} = 8.2 Hz), 146.6, 137.0, 129.9, 128.4, 127.9, 127.8,$ 125.4, 120.4 (d, $J_{CP} = 4.4$ Hz), 119.3 (d, $J_{CP} = 170.5$ Hz), 78.7 (d, $J_{\rm CP} = 8.0$ Hz), 73.8, 70.7 (d, $J_{\rm CP} = 10.7$ Hz), 68.0 (d, $J_{\rm CP} =$ 7.1 Hz), 38.1; ³¹P NMR δ 6.1. Exact mass: calculated for C₁₉H₂₂O₇PS (M+1) 425.0824; found 425.0809 (FAB).

(2S_P,5R,6R)-5,6-Dihydro-2-phenoxy-2H-1,2-oxaphosphorinane-5-azido-6-[(phenylmethoxy)methyl]-2-oxide [(-)-29]. Mesylated alcohol 28 (63 mg, 0.148 mmol) was taken up in DMF (0.3 mL) at room temperature, followed by the subsequent addition of 15-crown-5 ether (33 mg, 0.148 mmol) and sodium azide (9 mg, 0.140 mmol). Upon completion (2.5-3 h), removal of solvent under reduced pressure followed by flash chromatography (1:1 EtOAc/hexanes) afforded 45 mg (82%) of compound **29**; [α]_D -316.3 (*c* 0.4, CH₂Cl₂); IR (neat) 3062, 2106, 1590, 1488, 1271, 1198 cm $^{-1};$ $^1{\rm H}$ NMR δ 7.12–7.39 (m, 10H), 6.85 (ddd, $J_{\rm HP} = 46.4$, J = 12.5, 5.6 Hz, 1H), 6.44 (dd, J =17.4, J= 12.9 Hz, 1H), 4.93 (ddd, J= 11.4, 6.9, 3.3 Hz, 1H), 4.59 (s, 2H), 3.89 (m, 1H), 3.84 (d, J = 6.39 Hz, 2H); ¹³C NMR δ 149.5 ($J_{\rm CP} = 8.3$ Hz), 142.2, 137.3, 129.8 ($J_{\rm CP} = 11.0$ Hz), $128.6 (J_{\rm CP} = 10.1 \text{ Hz}), 128.1 (J_{\rm CP} = 5.7 \text{ Hz}), 127.8, 125.4, 121.8$ $(J_{\rm CP} = 171.2 \text{ Hz}), 120.4 (J_{\rm CP} = 4.5 \text{ Hz}), 78.3 (J_{\rm CP} = 7.1 \text{ Hz}),$ 73.9, 69.1 ($J_{\rm CP}$ = 9.3 Hz), 54.2 ($J_{\rm CP}$ = 11.1 Hz); ³¹P NMR δ 5.72. Exact mass: calcd for C₁₈H₁₉N₃O₄P (M+1) 372.1113; found 372.2313 (FAB)

(2S_P,3S,4R,5R,6R)-2-Phenoxy-6-[(phenylmethoxy)methyl]-1,2-oxaphosphorinane-3,4-diol-5-azido-2-oxide [(+)-30]. To a solution of 29 (12 mg, 0.032 mmol) in acetone (0.9 mL) and t-BuOH (0.3 mL) were added citric acid (6 mg, 0.031 mmol), NMO+H_2O (5 mg, 0.037 mmol), and OsO_4 (1 drop of a 4% solution in water). After stirring for 24 h at room temperature, one more drop OsO₄ was added, and the stirring was continued for 8 h. Water was added to the reaction mixture and extracted with EtOAc (3x). Removal of the solvent under reduced pressure and flash chromatography (3:2 EtOAc/ hexanes) afforded **30** (7 mg, 54%) as an oil; $[\alpha]_D$ +41.1 (c $0.0072, CH_2Cl_2$; IR (neat) 3343, 3065, 2922, 2113, 1590, 1491, 1454, 1256, 1201, 1024, 981, 938 cm⁻¹; ¹H NMR δ 7.38–7.11 (m, 10H), 5.07-5.05 (m, 1H), 4.56 (s, 2H), 4.50-4.39 (m, 2H), 4.06 (d, J=3.7 Hz, 1H), 3.78–3.70 (m, 2H); $^{13}\mathrm{C}$ NMR δ 150.0 (d, $J_{\rm CP} = 10.3$ Hz), 137.2, 129.7, 128.5, 127.9, 125.2, 120.4 (d, $J_{\rm CP} = 4.2$ Hz), 74.8 (d, $J_{\rm CP} = 6.8$ Hz), 73.6, 71.7 (d, $J_{\rm CP} = 3.2$ Hz), 68.1 (d, $J_{CP} = 11.0$ Hz), 64.3 (d, $J_{CP} = 146.7$ Hz), 61.6; ³¹P NMR δ 16.2. Exact mass: calcd for $C_{18}H_{21}N_3O_6P$ (M+1) 406.1168; found 406.1159 (FAB).

 $(2S_P, 3R, 4S, 5S, 6R)$ -2-Phenoxy-6-[(phenylmethoxy)methyl]-1,2-oxaphosphorinane-3,4,5-triol-5-2-oxide [(+)-**31].** To a solution of **27** (65 mg, 0.187 mmol) in acetone (5.2 mL) and tert-butyl alcohol (2.1 mL) were added citric acid (72 mg, 0.374 mmol), NMM (100 µl, 0.560 mmol), and a 1% solution of OsO_4 in water (24 μ L). A solution of *m*-CPBA (137 mg, 0.560 mmol) in acetone (1 mL) was added over 30 min. After 24 h of stirring, additional amounts of OsO4 (0.024 mL) and *m*-CPBA (70 mg) were added, and the reaction was stirred for an additional 24 h. The reaction was guenched with Na₂SO₃ (sat'd aq.) and extracted with CH₂Cl₂ (3x), and the combined organic layers were dried (Na $_2SO_4$). ³¹P analysis of the crude reaction mixture showed a diastereomeric ratio of 15:1. Subsequent flash chromatography (9:1 EtOAc/MeOH) afforded 38 mg (53%) of **31** as the major isomer; $[\alpha]_D$ +21.6 (*c*

736, 690 cm⁻¹; ¹H NMR δ 7.31–7.13 (m, 10H), 4.52–4.49 (m, 2H), 4.47–4.43 (m, 1H), 4.21–4.18 (m, 1H), 4.03 (dd, $J_{\rm HP}$ = 9.53, 9.53 Hz, 1H), 3.96 (dd, $J_{\rm HP}$ = 11.3, 1.81 Hz, 1H), 3.86–3.84 (m, 1H), 3.76 (dd, $J_{\rm HP}$ = 10.9, 5.81 Hz); ¹³C NMR δ 149.5 (d, $J_{\rm CP}$ = 9.6 Hz), 137.8, 130.0, 128.4, 127.6, 127.6, 125.5, 120.0 (d, $J_{\rm CP}$ = 4.4 Hz), 78.7 (d, $J_{\rm CP}$ = 3.6 Hz), 73.4, 73.3 (d, $J_{\rm CP}$ = 7.2 Hz), 69.8 (d, $J_{\rm CP}$ = 12.6 Hz), 66.8, 65.7 (d, $J_{\rm CP}$ = 145.9 Hz); ³¹P NMR δ 17.1. Exact mass: calcd for C₁₈H₂₂O₇P (M+1) 381.1025; found 381.1103 (FAB).

Trichlorooxazoline-Containing Phosphonate [(-)-32]. To a solution of $\boldsymbol{27}\,(32\text{ mg},\,0.092\text{ mmol})$ in $CH_2Cl_2\,(2\text{ mL})$ were added Cl₃CCN (0.01 mL) and a catalytic amount of DBU at -50 °C. The solution was allowed to warm to -20 °C over 5 h. The reaction mixture was quenched with NH₄Cl (sat'd aq.), extracted with CH₂Cl₂ (3x) and dried (Na₂SO₄). Removal of solvent under reduced pressure and flash chromatography (1:1 hexanes/EtOAc) afforded 36 mg (80%) of 32; $[\alpha]_D - 1.88$ (c 2.6, CH₂Cl₂); IR (neat) 3063, 3030, 2927, 2868, 1716, 1664, 1591, 1491 1454, 1271, 1203, 985, 930 cm⁻¹; ¹H NMR δ 7.38-7.15 (m, 10H), 5.12-5.08 (m, 1H), 4.99-4.89 (m, 1H), 4.70-4.57(m, 3 H), 3.92 (d, J = 11.2 Hz, 1H), 3.82 (dd, J = 11.1, 4.6 Hz, 1H), 2.75 (dd, J = 18.2, 15.9, 7.0 Hz, 1H), 2.32 (dd, J = 17.0, 7.5 Hz, 1H); ¹³C NMR δ 162.3, 149.5 (d, $J_{\rm CP}$ = 8.5 Hz), 137.1, 129.8, 128.5, 128.0, 127.8, 120.7 (d, $J_{\rm CP} = 4.3$ Hz), 85.8, 79.2 $(d, J_{CP} = 7.6 \text{ Hz}), 77.4, 73.8, 69.8 (d, J_{CP} = 5.9 \text{ Hz}), 63.1 (d, J_{CP})$ = 3.7 Hz), 24.9 (d, $J_{\rm CP}$ = 126.3 Hz); $^{31}{\rm P}$ NMR δ 19.5. Exact mass: calcd for $C_{20}H_{20}Cl_3NO_5P(M+1)$ 490.0145; found 490.0145 (FAB).

Trichloroacetimidate-Containing Phosphonate [(+)-33]. Compound 32 (15 mg, 0.031 mmol) was dissolved in acetone (0.6 mL), followed by the addition of H_2O (0.1 mL). PPTS (2 mg, 0.0093 mmols) was added, and the reaction was stirred at room-temperature overnight. Concentration under reduced pressure and flash chromatography (3:2 EtOAc/ hexanes), provided 12 mg (76% yield) of the **33** as an oil; $[\alpha]_D$ +36.0 (c 1.6, CH₂Cl₂); ¹H NMR δ 7.70 (d, J = 7.5 Hz, 1H), 7.37-7.19 (m, 10H), 4.17 (dddd, J = 33.6, 8.0, 8.0, 4.0 Hz, 1H), 4.62 (d, J = 11.8 Hz, 1H), 4.57 (d, J = 11.8 Hz, 1H), 4.48 (ddd, J = 8.8, 8.8, 4.4 Hz, 1H), 4.17 (dd, J = 9.0, 3.6 Hz, 1H), 3.85 (m, 2H), 2.61 (dd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9, 4.6 Hz, 1H), 2.41 (ddd, J = 15.9, 15.9 19.1, 15.9, 4.6 Hz, 1H); ¹³C NMR δ 163.2, 149.4 (d, $J_{\rm CP} = 9.3$ Hz), 137.2, 130.2, 128.6, 128.1, 127.8, 125.8, 120.0 (d, $J_{\rm CP}$ = 4.6 Hz), 95.2, 73.9, 72.2, 70.0 (d, $J_{\rm CP} = 8.0$ Hz), 68.5 (d, $J_{\rm CP} =$ 6.1 Hz), 49.4 (d, $J_{\rm CP}$ = 7.8 Hz), 25.3 (d, $J_{\rm CP}$ = 126.3 Hz); ³¹P NMR δ 18.7.

(1S,S_P)-2-Propenyl-1-triphenylmethoxymethyl-2-propenyl Phenylmethoxymethyl Ester Phosphonic Acid [(-)-34]. To a solution of benzyl alcohol (0.51 mL, 4.93 mmol) in dry DME (25 mL) was added BuLi (3.2 mL 1.3 M in hexanes) at -30 °C, and the solution was warmed to room temperature over 1 h and stirred for an additional 4 h. This solution was added to the mixed phosphonate (1.320 g, 2.59mmol) in DME (15 mL) at -50 °C over 1 h. The resulting mixture was transferred back to the alkoxide flask at -50 °C and allowed to warm to room temperature overnight. This mixture was stirred for 8 h at room temperature, quenched with NH₄Cl (sat'd aq.), extracted with EtOAc (3x), and dried (Na_2SO_4) . Removal of the solvent under reduced pressure and flash chromatography (7:3 hexanes/EtOAc) afforded 1.20 g (88%) of **34** as an oil; $[\alpha]_D - 4.6 (c \ 0.16, CH_2Cl_2)$; IR (neat) 3059, 3032, 2931, 1638, 1490, 1448, 1257, 1215, 1078, 990 cm⁻¹; ¹H NMR & 7.49-7.19 (m, 20H), 5.94-5.78 (m, 2H), 5.43-5.01 (m, 7H), 3.30–3.21 (m, 2H), 2.68 (dd, J = 21.1, 7.3 Hz, 2H); ¹³C NMR δ 143.6, 136.3 (d, $J_{\rm CP}$ = 6.9 Hz), 134.2 (d, $J_{\rm CP}$ = 3.1 Hz), 128.6, 128.3, 128.0, 127.7, 127.6, 127.3 (d, $J_{\rm CP} = 11.5$ Hz), 127.0, 120.0 (d, $J_{CP} = 14.8$ Hz), 118.4, 86.7, 76.7 (d, $J_{CP} = 15.6$ Hz), 66.9 (d, $J_{CP} = 6.4$ Hz), 66.3 (d, $J_{CP} = 5.2$ Hz), 32.6 (d, $J_{CP} =$ 139.6 Hz); ^{31}P NMR δ 28.3. Exact mass: calcd for $C_{33}H_{34}O_4P$ (M+1) 525.2116; found 525.2195 (FAB).

(2S_P,6S)-2-Phenylmethoxy-6-[(triphenylmethoxy)methyl]-2H-1,2-oxaphosphorin-3,6-dihydro-2-oxide, [(-)-35]. To a solution of the 34 (70 mg, 0.13 mmol) in dry CH₂Cl₂ (13 mL) was added catalyst A (3 mg, 3 mol %), and the mixture was refluxed for 6 h, diluted with CH₂Cl₂, flushed with air, and stirred overnight at room temperature. Removal of the solvent under reduced pressure and flash chromatography (7:3 EtOAc/hexanes) afforded 63 mg (98%) of **35** as white solid; mp 153–155 °C; [α]_D –38.4 (c 0.56, CH₂Cl₂); IR (neat) 3058, 3032, 2929, 1490, 1448, 1279, 1248, 1077, 1001, 899 cm⁻¹; ¹H NMR δ 7.45–7.21 (m, 20H), 5.82–5.72 (m, 2H), 5.19–5.14 (m, 2H), 4.19 (dd, J = 11.9, 8.3 Hz, 1H), 3.37 (dd, J = 9.5, 6.3 Hz, 1H), 3.26 (dd, J = 9.6, 5.3 Hz, 1H), 2.50–2.04 (m, 2H); ¹³C NMR δ 143.5, 136.0 (d, J_{CP} = 6.3 Hz), 128.6, 128.5, 128.3, 127.8, 127.1, 126.7 (d, J_{CP} = 16.9 Hz), 121.0 (d, J_{CP} = 9.5 Hz), 86.7, 78.8 (d, J_{CP} = 7.4 Hz), 67.1 (d, J_{CP} = 6.0 Hz), 65.6 (d, J_{CP} = 5.0 Hz), 22.3 (d, J_{CP} = 132.3 Hz); ³¹P NMR δ 21.4. Exact mass: calcd for C₃₁H₃₀O₄P (M+1) 497.1882; found 497.1883 (FAB).

(2Sp,4R,5S,6S)-2-Phenylmethoxy-6-[(triphenylmethoxy)methyl]-1,2-oxapho sphorinane-4,5-diol-2-oxide [(+)-36]. To a solution of $\mathbf{35}$ (61 mg, 0.12 mmol) in acetone (2 mL) and t-BuOH (0.8 mL) were added citric acid (30 mg, 0.16 mmol), NMO (18 mg, 0.13 mmol) and OsO4 (1 drop 4% solution in water). The reaction mixture was stirred for 48 h, solid Na₂SO₃ and 3% aq. NaHSO₄ were added, and the mixture was stirred for 30 min and extracted with EtOAc (3x). Removal of the solvent under reduced pressure and flash chromatography (EtOAc/4% MeOH) afforded 51 mg (81%) of **36** as an oil; $[\alpha]_D$ +26.0 (c 0.19, CH₂Cl₂); IR (neat) 3363, 3059, 3033, 2934, 1490, 1449, 1226, 1079, 1022, 983 cm⁻¹; ¹H NMR δ 7.44–7.19 (m, 20H), 5.13–5.01 (m, 2H), 4.70–4.64 (m, 1H), 4.26 (br. d, J =31.6 Hz, 1H), 3.89 (d, J = 7.8 Hz, 1H), 3.39 (d, J = 2.8 Hz, 2H), 2.31 (ddd, J = 15.6, 15.2, 5.7 Hz, 1H), 1.98 (ddd, J = 18.1, 14.9, 3.4 Hz, 1H); $^{13}\mathrm{C}$ NMR δ 143.6, 135.6 (d, J_{CP} = 5.8 Hz), 128.6, 128.5, 127.9, 128.8, 127.1, 86.7, 76.5 (dd, $J_{\rm CP}$ = 4.9 Hz), 68.8 (d, $J_{\rm CP} = 5.4$ Hz), 67.6 (d, $J_{\rm CP} = 6.4$ Hz), 67.3 (d, $J_{\rm CP} = 7.1$ Hz), 63.2 (d, $J_{\rm CP}$ = 6.4 Hz), 28.6 (d, $J_{\rm CP}$ = 124.6 Hz); ³¹P NMR δ 26.6.

(2Sp,4R,5S,6S)-2-Phenylmethoxy-6-[(triphenylmethoxy)methyl]-1,2-oxaphosphorinane-4,5-carbonate-2-oxide. To a solution of 36 (23 mg, 0.043 mmol) and triphosgene (27 mg, 0.09 mmol) in CH_2Cl_2 (1.0 mL) was added Et_3N (0.03 mL, 0.22 mmol) at $-40\,$ °C, and the solution was warmed to room temperature over 5 h. The reaction mixture was diluted with CH₂Cl₂, quenched with NH₄Cl (sat'd aq.), extracted with CH₂Cl₂, and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (3:2 EtOAc/ hexanes) afforded 20 mg (84%) of the carbonate as a white solid; mp 162–164 °C; $[\alpha]_{\rm D}$ +43.0 (c 3.6, CH₂Cl₂); IR (neat) 3059, 3032, 2929, 1817, 1599, 1491, 1448, 1354, 1272, 1170, 1069, 1034, 965 cm⁻¹; ¹H NMR & 7.46-7.26 (m, 20H), 5.25-5.19 (m, 2H), 5.07 (t, J = 8.7 Hz, 1H), 5.00 - 4.91 (m, 1H), 4.47 -4.44 (m, 1H), 3.63 (dd, J = 11.0, 2.3 Hz, 1H), 3.36 (dt, J =11.0, 3.0 Hz, 1H), 2.62-2.52 (m, 1H), 2.32 (ddd, J = 18.3, 15.8, 9.5 Hz, 1H); ¹³C NMR δ 152.5, 143.0, 135.2 (d, $J_{CP} = 5.6$ Hz), 128.9, 128.8, 128.4, 128.0, 127.9, 127.2, 86.9, 75.0 (d, $J_{\rm CP}=$ 5.7 Hz), 71.7, 71.1 (d, $J_{CP} = 6.5$ Hz), 68.3 (d, $J_{CP} = 6.1$ Hz), 61.9 (d, $J_{\rm CP}$ = 7.6 Hz), 24.7 (d, $J_{\rm CP}$ = 121.7 Hz); ³¹P NMR δ 21.6. Exact mass: calcd for C₃₂H₃₀O₈P (M+1) 557.1729; found 557.1705 (FAB).

(2S_P,5S,6R)-5,6-Dihydro-2-phenylmethoxy-6-[(triphenylmethoxy)methyl]2 H-1,2-oxaphosphorinane-5-ol-2-oxide [(+)-37]. To a solution of the carbonate (150 mg, 0.27 mmol) in THF (10.0 mL) was added KHMDS (0.71 mL 0.5 M in toluene, 0.35 mmol) at -40 °C, and the solution was warmed to 0 °C over 3 h. The reaction mixture was quenched with NH₄Cl (sat'd aq.), extracted with EtOAc (3x), and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (4:1 EtOAc/hexanes) afforded 110 mg (80%) of **37**; [α]_D +82.9 (c 1.64, CH₂Cl₂); IR (neat) 3325, 3058, 3032, 2930, 1597, 1490, 1448, 1362, 1228, 1193, 1079, 1019, 982 cm⁻¹; ¹H NMR δ 7.40–7.17 (m, 20H), 6.69 (ddd, J = 47.7, 12.9, 1.5 Hz, 1H), 5.72 (ddd, J = 18.8, 12.9, 2.2 Hz, 1H), 5.08–4.97 (m, 2H), 4.55–4.51 (m, 2H), 3.70 (br. s, 1H), 3.44 (d, J = 1.8 Hz, 2H); ¹³C NMR δ 153.1, 143.4, 135.6,128.5, 127.9, 127.8, 127.1, 115.7 (d, $J_{\rm CP}$ = 166.2 Hz), 86.8, 79.5 (d, $J_{\rm CP}$ = 5.6 Hz), 68.5 (d, $J_{\rm CP}$ = 5.7 Hz), 65.6 (d, $J_{\rm CP}$ = 10.0 Hz), 63.1 (d, $J_{\rm CP}$ = 7.7 Hz); ³¹P NMR δ 12.7. Exact mass: calcd for C₃₁H₃₀O₅P (M+1) 513.1831; found 513.1812 (FAB).

(2S_P,3S,4R,5S,6R)-2-Phenylmethoxy-6-[hydroxymethyl]-1,2-oxaphosphorinane-3,4,5-triol-2-oxide) (38). To a solution of OsO₄ (43 mg, 0.017 mmol) in CH₂Cl₂ (0.6 mL) was added TMEDA (22 mg, 0.19 mmol) followed by the starting alcohol (68 mg, 0.13 mmol) in CH_2Cl_2 (1.0 mL) and at -78 °C. The reaction mixture was stirred for 3 h at -78 °C, warmed to room temperature, and stirred for 15 min. Removal of the solvent under reduced pressure afforded the crude osmate ester, which was dissolved in 1.0 mL MeOH and treated with citric acid (40 mg, 0.21 mmol) for 24 h. The solvent was removed, the residue was dissolved in a small amount of MeOH, and the solvent was filtered through silica gel with EtOAc/10%MeOH. The crude product was dissolved in 1.0 mL MeOH and treated with catalytic amount of TsOH·H₂O for 8 h. Removal of the solvent under reduced pressure and flash chromatography (EtOAc/10% MeOH) afforded 28 mg (71%) of 38; ¹H NMR δ 7.44-7.31 (m, 5H), 5.21-5.11 (m, 2H), 4.55-4.50 (m, 1H), 4.24 (dt, J = 33.7, 2.7 Hz, 1H), 4.01 (dd, J = 9.8, 3.4 Hz, 1H), 3.90 (ddd, $J=12.5,\,4.4,\,2.9$ Hz, 1H), 3.75 (dd, J= 9.8, 2.1 Hz, 1H); $^{13}\mathrm{C}$ NMR δ 137.7 (d, J_{CP} = 6.4 Hz), 129,7, 129.6, 129.2, 78.4 (d, $J_{\rm CP}$ = 4.4 Hz), 75.5 (d), 71.4, 69.6 (dt, $J_{\rm CP}$ = 6.4 Hz), 69.0 (d), 67.7 (d, $J_{\rm CP}$ = 144.5 Hz), 62.8 (dt, $J_{\rm CP}$ = 8.0 Hz); ³¹P NMR δ 24.5. Exact mass: calcd for C₃₈H₃₆O₈P (M+1) 651.2148; found 651.2131 (FAB).

(2S_P,5S,6R)-5,6-Dihydro-2-phenylmethoxy-2H-1,2oxaphosphorinane-5-methanesulfonate-6-[(triphenylmethoxy)-methyl]-2-oxide [(+)-39]. To a solution of diol 36 (180 mg, 0.339 mmol) in CH₂Cl₂ (10 mL) were added a catalytic amount of DMAP (8.3 mg, 0.068 mmol), MsCl (0.087 mL, 1.13 mmol) at 0 °C followed by Et₃N (0.191 mL, 1.36 mmol). The reaction mixture was warmed to room temperature over 1 h and stirred overnight at room temperature. Et₃N (0.034 mL, 0.339 mmol) was added, and the stirring was continued for 6 h. After addition of brine the mixture was extracted with EtOAc (3x) and dried (Na₂SO₄). Removal of the solvent under reduced pressure and flash chromatography (1:1 hexanes/EtOAc) afforded 120 mg (60%) of **39**; $[\alpha]_D$ +105.2 (c 1.9, CH₂Cl₂); IR (neat) 3032, 2933, 1490, 1449, 1364, 1259, 1180, 985, 961, 853 cm⁻¹; ¹H NMR δ 7.44–7.24 (m, 20H), 6.81 (ddd, J = 45.9, 12.9, 2.1 Hz, 1H), 6.03 (ddd, J = 17.2, 12.9, 2.1 Hz, 1H), 5.64-5.60 (m, 1H), 5.20-5.18 (m, 2H), 4.75-4.71 (m, 1H), $3.70 \,(\mathrm{dd}, J = 11.0, 2.8 \,\mathrm{Hz}, 1\mathrm{H}), 3.30 \,(\mathrm{dt}, J = 11.0, 3.0 \,\mathrm{Hz}, 1\mathrm{H}),$ 2.65 (s, 3H); $^{13}\mathrm{C}$ NMR δ 145.9, 142.8, 135.4 (d, $J_{\mathrm{CP}} = 5.9$ Hz), 128.7, 128.69, 128.6, 128.0, 127.9, 127.4, 119.9 (d, $J_{\rm CP} = 165.8$ Hz), 87.1, 76.9 (d, $J_{CP} = 6.1$ Hz), 70.9 (d, $J_{CP} = 10.6$ Hz), 68.7 $(d, J_{CP} = 6.2 \text{ Hz}), 61.4 (dt, J_{CP} = 6.3 \text{ Hz}), 38.1; {}^{31}\text{P} \text{ NMR} \delta 10.9.$ Exact mass: calcd for C₃₂H₃₂O₇PS (M+1) 591.1606; found 591.1591 (FAB).

(2S,5R,6R)-5,6-Dihydro-2-phenylmethoxy-2H-1,2oxaphosphorinane-5-ol-b enzoate-6-[(triphenylmethoxy)methyl]-2-oxide, [(-)-40]. To a solution of 39 (24 mg, 0.044 mmol) in DMF (0.5 mL) was added KOBz (10 mg, 0.062 mmol) and 18-Crown-6 (27 mg, 10 mmol). The resulting reaction mixture was stirred overnight at room temperature, quenched with NH₄Cl (sat'd aq.), extracted with EtOAc, and dried (Na_2SO_4) . Removal of the solvent under reduced pressure and flash chromatography (7:3 hexanes/EtOAc) afforded 20 mg (74%) of **40**; $[\alpha]_{\rm D} - 100.0 (c \ 0.51, \ CH_2Cl_2)$; ¹H NMR δ 7.82 (dd, J = 8.2, 1.2 Hz, 2H), 7.61 (t, J = 1.2 Hz, 1H), 7.52-7.12 (m, 22H), 7.03 (ddd, J = 45.5, 12.5, 5.7 Hz, 1H), 6.12 (dd, J = 17.2, 12.5 Hz, 1H), 5.72-5.69 (m, 1H), 5.15-4.97 (m, 3H), 3.56 (dd, J = 8.7, 5.5 Hz, 1H), 3.40 (t, J = 8.8 Hz, 1H); ¹³C NMR δ 165.0, 143.0, 142.8, 135.5 (d, $J_{\rm CP} = 6.2$ Hz), 133.4, 129.8, 129.0, 128.53, 128.48, 128.45, 128.3, 127.8, 127.6, 127.1, 122.5 (d, $J_{\rm CP}$ = 164.1 Hz), 87.0, 75.9 (d, J_{CP} = 5.1 Hz), 68.7 (dt, J_{CP} = 6.1 Hz), 63.9 (d, $J_{\rm CP}$ = 10.9 Hz), 61.0 (dt, $J_{\rm CP}$ = 10.5 Hz); ³¹P NMR δ 10.3. Exact mass: calcd for $C_{38}H_{34}O_6P\,(M+1)$ 617.2093; found 617.2082 (FAB).

(2S_P,5R,6R)-5,6-Dihydro-2-phenylmethoxy-2H-1,2oxaphosphorinane-5-ol-6 [(triphenylmethoxy)methyl]-2-oxide [(+)-41]. To a solution of 40 (40 mg, 0.065 mmol) in MeOH (1.0 mL) was added Mg $(OMe)_2 (0.06 \text{ mL} 7-8\% \text{ solution})$ in MeOH) the mixture was stirred overnight at room temperature, quenched with NH₄Cl (sat'd aq.). MeOH was removed, the residue extracted with EtOAc (3x), and dried (Na_2SO_4) . Removal of the solvent under reduced pressure and flash chromatography (7:3 EtOAc/hexanes) afforded 25 mg (75%) of **41**; $[\alpha]_D$ +1.27 (c 0.63, CH₂Cl₂); ¹H NMR δ 7.45-7.24 (m, 20H), 6.82 (ddd, *J* = 45.9, 12.5, 5.5 Hz, 1H), 5.09 (dd, *J* = 17.7, 12.5 Hz, 1H), 5.16-5.02 (m, 2H), 4.79-4.75 (m, 1H), 4.28 (br.s, 1H), 3.56-3.45 (m, 2H); ¹³C NMR δ 146.1, 143.1, 135.8 (d, J_{CP} = 6.0 Hz), 128.5, 128.4, 128.1, 127.9, 127.3, 119.9 (d, $J_{\rm CP}$ = 165.5 Hz), 87.3, 76.9 (d, $J_{\rm CP}$ = 4.9 Hz), 68.7 (dt, $J_{\rm CP}$ = 6.1 Hz), 63.4 (d, $J_{\rm CP}$ = 10.4 Hz), 62.7 (dt, $J_{\rm CP}$ = 9.6 Hz); $^{31}{\rm P}$ NMR δ 12.1.

(2S_P,3S,4R,5R,6R)-2-Phenylmethoxy-6-[(triphenylmethoxy)methyl]-1,2-oxaphosphorinane-3,4-diol-5-ol-benzoate-2-oxide) [(-)-42]. To a solution of 40 (33 mg, 0.054 mmol) in acetone (1.5 mL) and t-BuOH (0.5 mL) were added citric acid (13 mg, 0.068 mmol), NMO·H₂O (9 mg, 0.067 mmol) and OsO₄ (1 drop 4% solution in water). After stirring for 24 h at room temperature, one more drop of OsO4 was added, and the stirring was continued for 24 h. Water was added to the reaction mixture and extracted with EtOAc (3x). Removal of the solvent under reduced pressure and flash chromatography (EtOAc/1% MeOH) afforded 28 mg (80%) of 42 as an oil; $[\alpha]_D$ -36.4 (c 1.3, CH₂Cl₂); IR (neat) 3325, 3061, 3033, 2926, 1728, 1600, 1491, 1449, 1264, 1241, 1093, 999, 967 cm $^{-1};$ $^1\mathrm{H}$ NMR δ 7.81-7.80 (m, 2H), 7.60-7.56 (m, 1H), 7.40-7.29 (m, 13H), 7.16-7.08 (m, 9H), 5.74 (d, J = 4.5 Hz, 1H), 5.38 (dd, J = 6.5,6.5 Hz, 1H), 5.20-5.17 (m, 2H), 4.68 (ddd, J = 33.5, 3.8, 3.8Hz, 1H), 4.19 (dd, J = 10.7, 2.7 Hz, 1H), 3.46 (dd, J = 8.5, 6.0 Hz, 1H), 3.22 (dd, J = 8.6, 8.6 Hz, 1H); $^{13}\mathrm{C}$ NMR δ 164.5, 143.1, $135.5 (d, J_{CP} = 6.0 Hz), 133.4, 129.8, 129.0, 128.6, 128.5, 128.4,$ 128.0, 127.8, 127.0, 86.9, 74.6 (d, $J_{\rm CP}$ = 4.7 Hz), 71.5, 71.4, 68.9 $(dt, J_{CP} = 6.1 \text{ Hz}), 66.6 (d, J_{CP} = 140.9 \text{ Hz}), 60.7 (dt, J_{CP} = 11.6$ Hz); ³¹P NMR δ 24.3. Exact mass: calcd for C₃₈H₃₆O₈P (M+1) 651.2148; found 651.2131 (FAB).

(2S_P,3S,4R,5R,6R)-2-Phenylmethoxy-6-[hydroxymethyl]-1,2-oxaphosphorinane-3,4-diol-5-ol-2-oxide (43). To a solution of 42 (24 mg, 0.047 mmol) in acetone (0.9 mL) and t-BuOH (0.3 mL) were added citric acid (10 mg, 0.052 mmol), NMO· H₂O (7 mg, 0.052 mmol) and OsO₄ (1 drop 4% solution in water). The reaction was stirred for 24 h. Upon completion water was added to the reaction mixture and extracted with EtOAc (3x). The crude mixture was taken up in MeOH (1 mL), followed by the addition of catalytic TsOH, and stirred for an additional 24 h. Removal of the solvent under reduced pressure and flash chromatography (EtOAc/1% MeOH) afforded 10 mg (70%) of 43 as an oil; IR (neat) 3350, 3025, 2970, 1211, 1045 cm⁻¹; ¹H NMR δ 7.31–7.44 (m, 5H), 5.10–5.49 (m, 2H), 4.77 (dt, J = 5.9, 2.4 Hz, 1H), 4.29 (dd, J = 9.7, 3.7 Hz, 1H), 4.22(ddd, J = 33.4, 4.0, 4.0 Hz, 1H), 3.96 (d, J = 4.4 Hz, 1H), 3.76-3.84 (m, 2H); $^{13}\mathrm{C}$ NMR δ 136.7, 128.6, 128.4, 127.9, 77.6 (d, $J_{\rm CP} = 5.4$ Hz), 70.5 (d, $J_{\rm CP} = 2.3$ Hz), 68.5 (d, $J_{\rm CP} = 6.6$ Hz), 64.5 (d, $J_{\rm CP}$ = 144.9 Hz), 61.5 (d, $J_{\rm CP}$ = 10.1 Hz); ³¹P NMR δ 20.8.

Acknowledgment. This investigation was generously supported by funds provided by the National Institutes of Health (National Institute of General Medical Sciences, RO1-GM58103) and the NIH Dynamic Aspects in Chemical Biology Training Grant (A.W.). The authors thank Dr. David Vander Velde and Sarah Neuenswander for the assistance with the NMR measurements and Dr. Todd Williams for HRMS analysis.

P-Heterocyclic Building Blocks

The authors thank the National Science Foundation (Grant CHE-0079282) and the University of Kansas for funds to purchase the X-ray instrument and computers. The X-ray structure of compound **23** was determined by Dr. Douglas R. Powell. The authors kindly acknowledge Daiso Co., Ltd., Fine Chemical Department for donating 100 g of each antipode of both the benzyl and trityl protected glycidols (e-mail:akkimura@daiso.co.jp) and Materia, Inc. for supplying metathesis catalyst and helpful suggestions.

Supporting Information Available: We have previously published on compounds 15–17a.⁹ ¹H, ¹³C, and ³¹P NMR spectra for compounds 17b and 18–42, and X-ray crystallographic data for compound 23. This material is available free of charge via the Internet at http://pubs.acs.org.

JO0505122